

Available online at www.sciencedirect.com

Optical Materials xxx (2006) xxx–xxx

www.elsevier.com/locate/optmat

A new class of push–pull molecules for molecular electronics

Aldilene Saraiva-Souza ^a, Celso P. de Melo ^b, Paulo Peixoto ^b, Jordan Del Nero ^{a,*}

^a Departamento de Física, Universidade Federal do Pará (UFPA), 66075-110 Belém, PA, Brazil

^b Departamento de Física, Universidade Federal de Pernambuco (UFPE), 50670-901 Recife, PE, Brazil

Received 16 September 2005; received in revised form 23 March 2006; accepted 24 March 2006

8 Abstract

9 In this work, we have found that large conjugated pyridinium betaines, molecules composed by a donor (D) and acceptor (A) pair
 10 connected through a long conjugated chain, exhibit an inversion in the expected direction of the electron transfer resulting from photo-
 11 excitation. Once the connecting bridge is increased beyond a given size, the electron transfer is predicted to occur in a reversed manner,
 12 i.e. from the acceptor to the donor side of the molecule. This effect seems to be associated to an inversion of the spatial localization of the
 13 frontier molecular orbitals. We present results for the spatial localization of the frontier molecular orbitals and the theoretical absorption
 14 spectra of members of increasing size of a specific family of pyridinium betaines.

15 © 2006 Published by Elsevier B.V.

16 *Keywords:* Betaines; Semiempirical/CI methods; Absorption spectra

18 1. Introduction

19 By betaine, one designates not only the $[(\text{CH}_3)_3\text{N}^+\text{CH}_2$
 20 $\text{COO}^-]$ molecule (a naturally occurring zwitterion), but
 21 also several quaternary ammonium salts analogous in
 22 structure. Pyridinium betaines are good candidates for
 23 the design of photoactive molecular assemblies and the
 24 preparation of LB films with intense NLO response.

25 As we known, biological activity is directly correlate to
 26 molecular shapes [1,2], such as size and electronic transi-
 27 tions. Also, it is well-known [3] that the correlation of
 28 biological molecules can be strongly dependent on the elec-
 29 trostatic fields generated in the process of charge transfer.
 30 Those fields are determined by structural molecular factors
 31 [4]. Since measurements of conformation data are not yet
 32 available for this molecule, we will have to rely on simula-
 33 tion information gathered from electronic structure
 34 calculations.

35 Small pyridinium betaines have large molecular hyper-
 36 polarizabilities relative to their molecular sizes [5]. We have

examined the electronic and nonlinear optical properties of 37
 several families of betaine compounds [6]. 38

In this communication, we report the most important 39
 results found for one of these families (Fig. 1), where a 40
 given pair of D and A groups is connected through conju- 41
 gated bridges of increasing sizes. 42

2. Methodology 43

44 Since experimental data are not available for the geom-
 45 etries of the molecules investigated (Fig. 1), systematic
 46 geometry optimizations were required. Sophisticated semi-
 47 empirical methods such as AM1 (Austin method 1), and
 48 PM3 (parametric method 3) provide a good balance
 49 between quality and computational effort to treat organic
 50 compounds.

51 While geometric and basic electronic features, such as
 52 bond lengths, bond and dihedral angles, heats of forma-
 53 tion, dipole moment values, etc., are well described at
 54 AM1 and PM3 levels, excitation electronic energies are
 55 usually overestimated, as expected from zero differential
 56 overlap (ZDO) methods that do not include configuration
 57 interaction (CI) corrections.

* Corresponding author. Tel.: +55 91 3201 7423; fax: +55 91 3201 7403.
 E-mail address: jordan@ufpa.br (J.D. Nero).

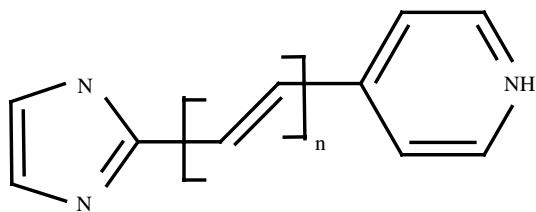


Fig. 1. Members of the family of pyridinium betaines investigated had an increasing number n of double bonds in the connecting conjugated bridge.

58 After the determination of optimized geometry, the
 59 absorption spectrum of each molecule was simulated by
 60 the INDO/S-CI (intermediate neglect of differential over-
 61 lap/spectroscopic-configuration interaction) [7–10], with
 62 parameters chosen to give the best description of the UV-
 63 NIS-Visible optical transitions [11–14] and 121 configura-
 64 tions were investigated for each molecule, including singlet
 65 states. We have taken account from the first 10-UMOs
 66 (unoccupied molecular orbitals) to the last 10-OMOs
 67 (occupied molecular orbitals) and adopted the Mattaga-
 68 Nishimoto γ 's.

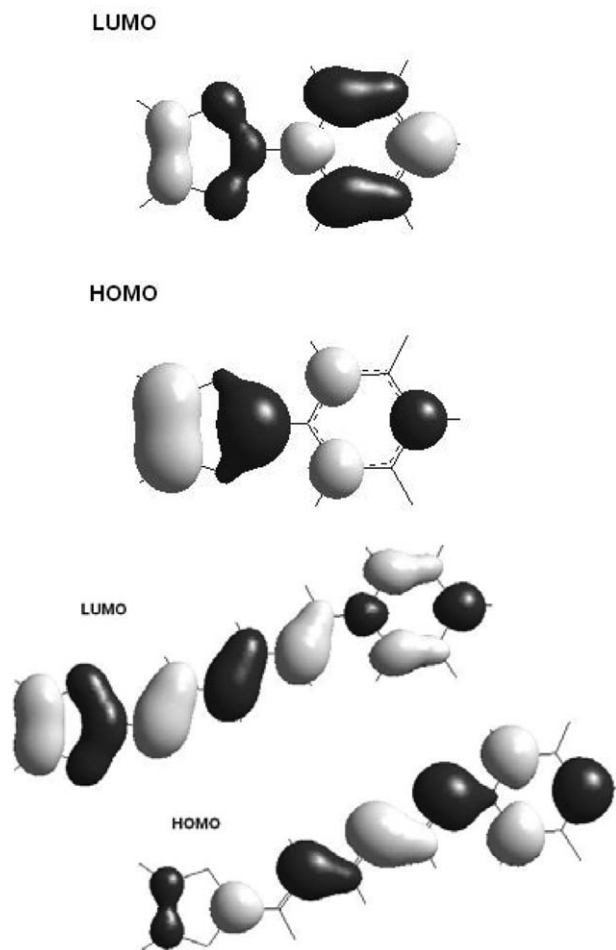


Fig. 2. Schematic representation of the calculated HOMO and LUMO for the $n = 0, 1, 2, 3, 4$ and 5 members of the betaine family considered.

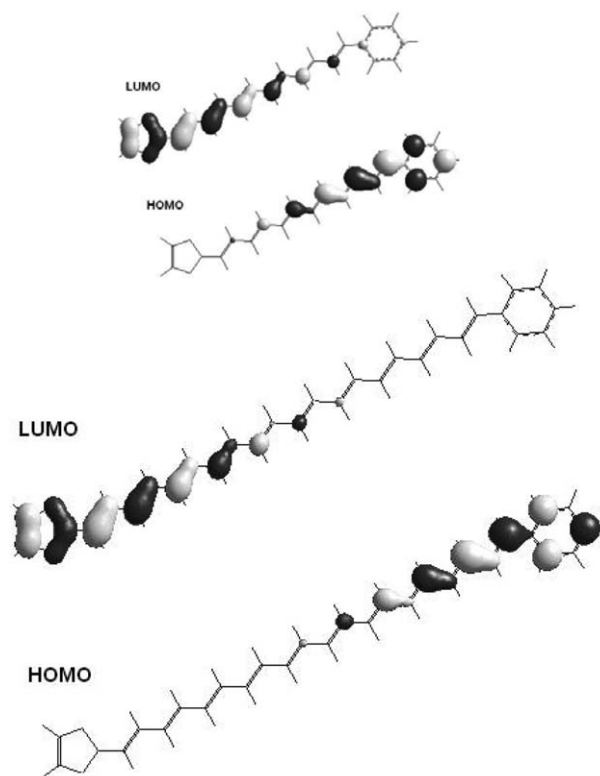


Fig. 3. Schematic representation of the calculated HOMO and LUMO for the $n = 6, 7, 8, 9$ and 10 members of the betaine family considered.

Also, we adopted well-known methodology for the prediction of the nonlinear polarizabilities (hyperpolarizability) presented elsewhere [15,16].

3. Results and discussion

Preliminary results in the investigation of the electronic structure of several pyridinium betaines [2] using semiempirical quantum chemical methods has revealed a very peculiar reversal of the charge transfer for the large molecules of each family. In this communication, we will show that this unusual behavior seems to be associated to a progressive change in the spatial localization of the frontier molecular orbitals (FMO) of the molecules that follows the increase in the size of the conjugated bridge connecting the D and A groups. For the smaller molecules of each family, the HOMO [LUMO] is more localized in the extremity of the molecule containing the D [A] group. In Fig. 2 we show the HOMO and LUMO representations for the smallest member (i.e. when the D and A groups are directly attached) and for $n = 2$ of the family represented in Fig. 1.

As it is shown in Fig. 3 for the $n = 6$ and $n = 10$ molecules, a definite inversion of the spatial localization of the HOMO and LUMO takes place beyond a certain length of the conjugated bridge (It is possible to see the inversion process on the reference.¹ As a consequence, one can expect

¹ The supplementary file has more FMO's.

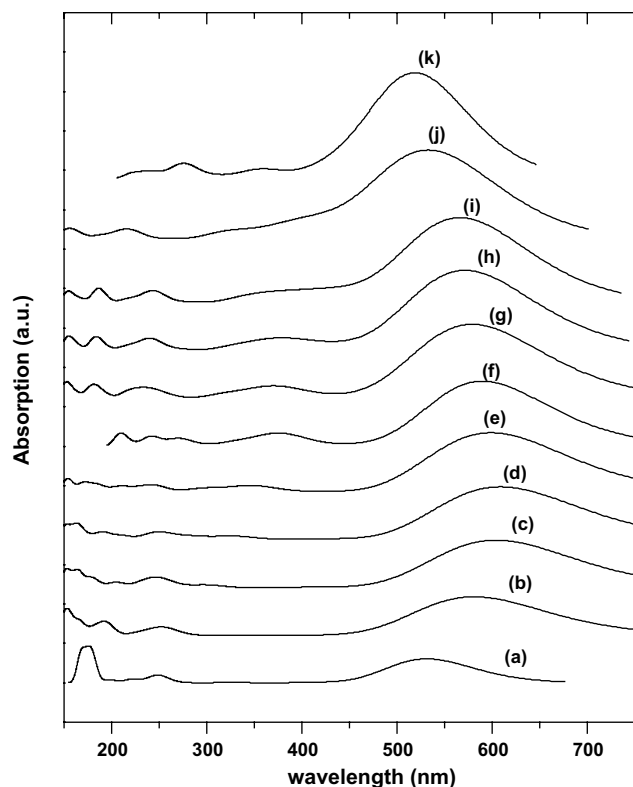


Fig. 4. Full INDO/S-CI theoretical absorption spectra of the $n = 0$ (a), $n = 1$ (b), $n = 2$ (c), $n = 3$ (d), $n = 4$ (e), $n = 5$ (f), $n = 6$ (g), $n = 7$ (h), $n = 8$ (i), $n = 9$ (j) and $n = 10$ (k), members of the betaine family considered calculated for the optimized geometries.

94 several unusual properties to be associated to these mole-
 95 cules, such as, for instance, a reversed photoinduced elec-
 96 tron transfer, i.e. following light excitation, charge will be
 97 transferred from the acceptor to the donor group.
 98 In Fig. 4 we plot the INDO/S-CI theoretical (gas phase)
 99 absorption spectra of some members of the family of
 100 betaines investigated. For the smallest molecule ($n = 0$)

Table 1
 Principal electronic UV-NIS-Vis absorption transitions as determined from INDO/S-CI calculations based on the optimized ground-state geometry from $n = 0$ to $n = 10$ members of the betaine family considered

| # // | o.s. (a.u.) | Absorption transitions |
|------|-------------|------------------------|
| 0 | 0.76 | 0.99 H → L⟩ |
| 1 | 3.94 | 0.69 H → L⟩ |
| 2 | 1.90 | 0.68 H → L⟩ |
| 3 | 2.14 | -0.68 H → L⟩ |
| 4 | 2.41 | 0.67 H → L⟩ |
| 5 | 2.69 | 0.63 H → L⟩ |
| 6 | 2.96 | -0.56 H → L⟩ |
| | | 0.34 H → L + 1⟩ |
| 7 | 3.22 | 0.48 H → L⟩ |
| | | 0.44 H → L + 1⟩ |
| 8 | 3.42 | 0.40 H → L⟩ |
| | | 0.52 H → L + 1⟩ |
| 9 | 3.58 | -0.32 H → L⟩ |
| | | -0.57 H → L + 1⟩ |
| 10 | 3.71 | -0.60 H → L + 1⟩ |

(Fig. 4a), we predict the existence of three main bands centered at ≈ 530 nm, 250 nm and 175 nm. For the member with $n = 5$ (Fig. 4f), we estimate that there will occur an increase in the intensity of the first band, while the second one is split in two. Finally, for $n = 10$, i.e. after the inversion of the FMO has completely taken place, a blue shift of the main band is found (Fig. 4k). Details of the main predicted transitions are shown in Table 1.

In Table 1 we have the main transition for all molecules. It is possible to see the inclusion of HOMO to LUMO + 1 for $n = 6$ to $n = 9$. It gives two very important pattern: (i) Red shift on absorption spectra from $n = 0$ to $n = 5$ (from Fig. 4a–f) and main transition is HOMO to LUMO; (ii)

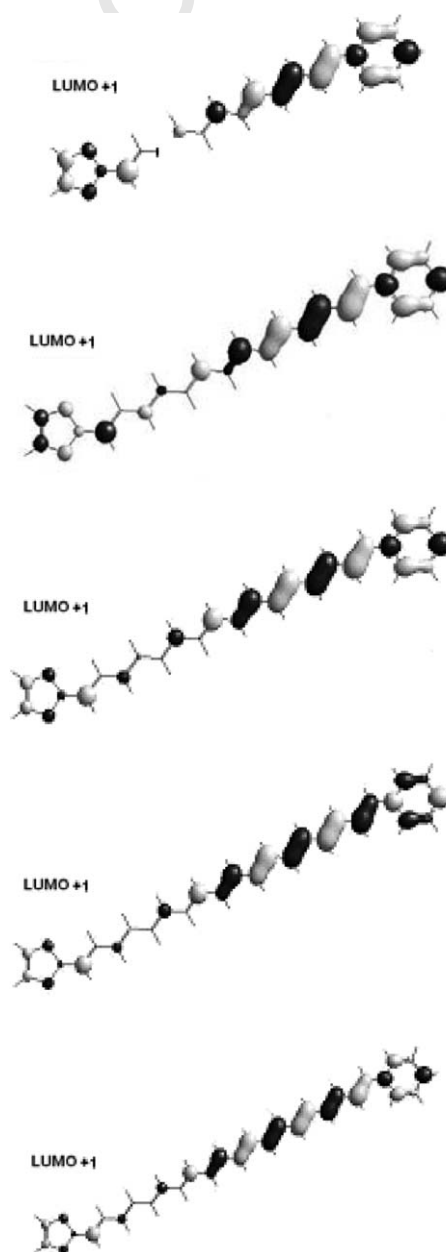


Fig. 5. Schematic representation of the calculated LUMO+1 for the $n = 6, 7, 8, 9$ and 10 members of the betaine family considered.

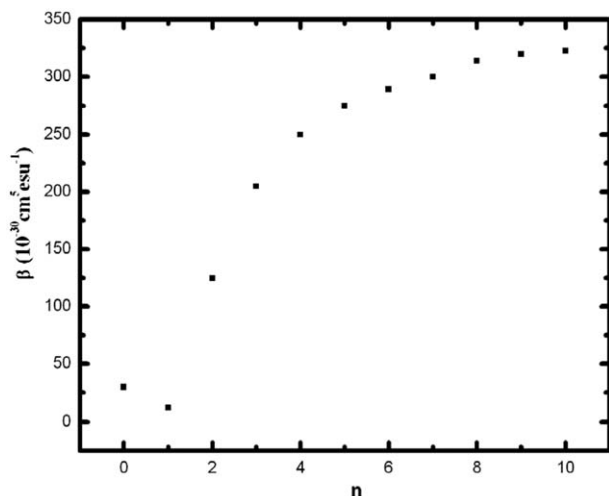


Fig. 6. Hyperpolarizability of betaine family considered.

114 Blue shift on absorption spectra from $n = 6$ to $n = 9$ (from
115 Fig. 4g–j) and main transition is composed by two species
116 (HOMO to LUMO and HOMO to LUMO + 1). Finally
117 only HOMO to LUMO + 1 for $n = 10$.

118 To clarify above results, in Fig. 5 it is possible to see
119 LUMO + 1 for $n = 6, 7, 8, 9, 10$. These results have similar
120 behavior with LUMO ($1 < n < 6$).

121 It could be expected a change of behavior on hyperpo-
122 larizability but in Fig. 6 it is presented and a canonical pat-
123 tern were founded.

124 To exclude the possibility of contamination of the
125 results as an artifact of the quantum chemical method used,
126 we have repeated these calculations using different method-
127 ologies. The existence of the predicted reversal in the spa-
128 tial localization of these FMO was confirmed in all cases.

129 4. Conclusions

130 As an example of a new effect predicted to occur in large
131 pyridinium betaines we present results for a specific family
132 of these compounds. For the largest molecules we have
133 found that there is in fact an inversion of the spatial local-

ization of the FMO. This is expected to have important
consequences not only on the optical, but also on several
other electronic properties of these betaines, which consti-
tute a new class of push–pull systems. This work is part of a
more complete investigation of the ground and excited
states of betaines to be reported elsewhere and at the
moment we are doing simulations applying external electric
field.

Acknowledgments

We thank the UFPA agency PROINT, FINEP, CAPES
and CNPq for financial support. ASS acknowledges a
CNPq fellowship.

Appendix A. Supplementary data

Supplementary data associated with this article can be
found, in the online version, at [doi:10.1016/j.optmat.2006.03.035](https://doi.org/10.1016/j.optmat.2006.03.035).

References

- [1] J.B.L. Pecq, Methods of Biochemical Analysis, Interscience, New York, 1971.
- [2] R.S. Braga, P.M.V.B. Barone, D.S. Galvao, Braz. J. Phys. 30 (2000) 560.
- [3] P.M.V.B. Barone, A. Camilo Jr., D.S. Galvao, Phys. Rev. Lett. 77 (1996) 1186.
- [4] J.A. Tuszynski, D.K. Basu, M. Kurzynski (Eds.), Introduction to Molecular Biophysics, CRC Press, New York, 2003.
- [5] J. Abe, Y. Shirai, J. Am. Chem. Soc. 118 (1996) 4705.
- [6] P.H.R. Peixoto, M.Sc. thesis, UFPE (Brazil), 2002.
- [7] J.D. Head, M.C. Zerner, Chem. Phys. Lett. 122 (1985) 264.
- [8] J.D. Head, M.C. Zerner, Chem. Phys. Lett. 131 (1986) 359.
- [9] W.P. Anderson, W.D. Edwards, M.C. Zerner, Inorg. Chem. 25 (1986) 2728.
- [10] W.D. Edwards, M.C. Zerner, Theoret. Chim. Acta 72 (1987) 347.
- [11] J. Del Nero, B. Laks, Synth. Met. 101 (1999) 440.
- [12] J. Del Nero, B. Laks, Synth. Met. 101 (1999) 379.
- [13] R.L. Doretto, J. Del Nero, B. Laks, Synth. Met. 101 (1999) 178.
- [14] J. Del Nero, C.P. de Melo, Synth. Met. 121 (2001) 1741.
- [15] A.G. Bezerra Jr., A.S.L. Gomes, D.A. da Silva-Filho, L.H. Acioli, C.B. de Araujo, C.P. de Melo, J. Chem. Phys. 111 (1999) 5102.
- [16] B. Kirtman, B. Champagne, Int. Rev. Phys. Chem. 16 (1997) 389.